ACTIVE CARBON ADSORPTION EFFICIENCY

Genano air purifiers utilize a high-quality active carbon with a highly porous structure. Each gram of active carbon has a surface area of approx. 900 m².





For instructions on changing the active carbon collector, please see the unit service manuals. This document withholds information about the adsorption efficiency of the active carbon.

HOW ACTIVE CARBON WORKS

Vapour-phase impurities are removed in three steps:

- 1. Transport of contaminant to the surface of active carbon;
- 2. Diffusion of the contaminant into the structure; and
- 3. Attraction of the contaminant to an interior surface by van der Waals forces.

Contaminant removal efficiency is dependent on many factors. The removal efficiency generally improves with increasing molecular mass, boiling point and concentration of the contaminant.

ADSORPTION EFFICIENCY FOR VARIOUS CHEMICAL COMPOUNDS

Capacity index (noncommittal data):

- **4 Good adsorption:** High capacity for compounds in this category. Active carbon takes up between 20 and 50% of its dead weight in gaseous materials. Average separating ability for gaseous materials in this category is approx. 35% of the dead weight of the active carbon. This category contains most odour-causing (smelly) substances.
- **3 Satisfactory adsorption:** Satisfactory capacity for compounds in this category. Active carbon takes up between 10 and 20% of its dead weight. Average receptiveness for gaseous materials in this category is approx. 15% of the dead weight of the active carbon.
- **2 Still considerable adsorption:** Includes compounds which are not highly adsorbed but may be taken up sufficiently to give satisfactory results in the operating conditions.
- **1 Very low adsorption:** The adsorption is low for these compounds. Active carbon cannot be satisfactorily used to remove them in ordinary circumstances.



| Compound | Molecular formula | Capacity index | Synonyms |
|--|---|----------------|--|
| 1-Pentanol | C ₅ H ₁₂ O | 4 | n-Pentanol |
| 2-Chlorobuta-1,3-diene | C ₄ H ₅ Cl | 4 | Chloroprene, 2-Chloro-1,3-butadiene, Chlorobutadiene |
| Acetaldehyde | C ₂ H ₄ O | 1 | Acetic aldehyde, ethanal, ethyl aldehyde |
| Acetic acid | C ₂ H ₄ O ₂ | 4 | Ethanoic acid, methanecarboxylic acid |
| Acetic anhydride | C ₄ H ₆ O ₃ | 4 | Acetic acid anhydride, Acetic oxide, Acetyl oxide |
| Acetone | C ₃ H ₆ O | 3 | Dimethyl ketone, ketone propane, 2-propanone |
| Acrolein | C ₃ H ₄ O | 2 | Acraldehyde, acrylaldehyde, acrylic aldehyde, allyl aldehyde, propenal, 2-propenal |
| Acrylic acid | C ₃ H ₄ O ₂ | 4 | Acroleic acid, ethylenecarboxylic acid, propenoic acid |
| Acrylonitrile | C ₃ H ₃ N | 4 | Acrylonitrile monomer, cyanoethylene, propenenitrile, vinyl cyanide |
| Amine | | 1 | |
| Ammonia | NH ₃ | 1 | Anhydrous ammonia |
| Amyl ether | C ₁₀ H ₂₂ O | 4 | Pentyl ether, amyl ether, diamyl ether |
| Aniline | C ₆ H ₇ N | 4 | Aminobenzene, aniline oil, benzamine, phenylamine |
| Benzene | C ₆ H ₆ | 4 | Benzol, phenyl hydride |
| Bromine | Br ₂ | 4 | |
| Butadiene | C ₄ H ₆ | 3 | Biethylene, bivinyl, divinyl, erythrene, vinylethylene |
| Butanal | C ₄ H ₈ O | 1 | Butyraldehyde |
| Butane | C ₄ H ₁₀ | 2 | Butyl hydride, methylethylmetane |
| Butanoic acid | C ₄ H ₈ O ₂ | 4 | Butyric acid, propanecarboxylic acid |
| Butanone | C ₄ H ₈ O | 4 | Ethyl methyl ketone, methyl acetone, methyl ethyl ketone, MEK, methylpropanone |
| Butyl acetate | C ₆ H ₁₂ O ₂ | 4 | Butyl ethanoate, butile, acetic acid n-butyl ester |
| Butyl alcohol | C ₄ H ₁₀ O | 4 | Butanol |
| Butyl chloride | C ₄ H ₉ Cl | 4 | |
| Butyl ether | C ₄ H ₈ O | 4 | |
| Butyl ethyl ketone | C ₇ H ₁₄ O | 4 | 3-heptanone, ethyl-butyl ketone, EBK |
| Camphor | C ₁₀ H ₁₆ O | 4 | 2-camphanone, formosa, 2-bornanone |
| Caprylic acid | C ₈ H ₁₆ O ₂ | 4 | Octanoic acid |
| Carbon dioxide | CO ₂ | 1 | Carbon acid gas, dry ice |
| Carbon disulphide | CS ₂ | 4 | Carbon bisulfide, methanedithione |
| Carbon monoxide | СО | 1 | Carbon oxide, monoxide |
| Chlorine | Cl ₂ | 2 | |
| Chlorobenzene | C ₆ H ₅ CI | 4 | Benzene chloride, phenyl chloride, chlorobenzol |
| Chloroethylene | C ₂ H ₃ Cl | 3 | Vinyl chloride, vinyl chloride monomer, VCM, chloroethene |
| Chloroform | CHCl ₃ | 4 | Trichloromethane, methyl trichloride |
| Creatine | C ₄ H ₉ N ₃ O ₂ | 4 | Methylguanidoacetic acid |
| Cresol | C ₇ H ₈ O | 4 | Hydroxytoluene, cresylic acid, hydroxymethylbenzene |
| Cyclohexane | C ₆ H ₁₂ | 4 | Hexanaphthene |
| Cyclohexanol | C ₆ H ₁₂ O | 4 | Cyclohexyl alcohol, hexahydrophenol, hydrophenol, hexalin, hydralin |
| Cyclohexene | C ₆ H ₁₀ | 4 | Tetrahydrobenzene, benzenetetrahydride, hexanaphthylene |
| Decane | C ₁₀ H ₂₂ | 4 | -1 · / · · |
| Dibromomethane | CH ₂ Br ₂ | 4 | Methyl dibromide, methylene dibromide |
| Dichlorobenzene | C ₆ H ₄ Cl ₂ | 4 | DCB |
| Dichlorodifluoromethane | CCl ₂ F ₂ | 4 | Carbon dichloride difluoride, difluorodichloromethane, Freon 12, CFC-12 |
| Dichloroethane | C ₂ H ₄ Cl ₂ | 4 | DCE |
| Dichlorodiethyl ether | C ₄ H ₈ Cl ₂ O | 4 | Bis(2-chloroethyl) ether, Chlorex, DCEE, Chloroethyl ether, Dichloroethyl ether |
| Dichloroethylene | C ₂ H ₂ Cl ₂ | 4 | Dichloroethene, vinylidene chloride |
| Dichlorofluoromethane | CHCl ₂ F | 3 | Fluorodichloromethane, Freon 21 |
| Dichloronitroethane | | | i idorodioniorometrane, Fleori Z I |
| | C ₂ H ₃ O ₂ NCl ₂ | 4 | Dranylana diablarida |
| Dichloropropane Dichlorototrofluoroothono | C ₃ H ₆ Cl ₂ | 4 | Propylene dichloride |
| Dichlorotetrafluoroethane | C ₂ Cl ₂ F ₄ | 4 | Freon 114 |
| Diethyl ether | C ₄ H ₁₀ O | 3 | Ethyl ether, ethoxyethane |
| Diethyl ketone | C ₅ H ₁₀ O | 4 | |
| Diethylamine | C ₄ H ₁₁ N | 2 | Diethamine, ethylethanamine |



| Compound | Molecular formula | Capacity index | Synonyms |
|--------------------------------------|---|----------------|---|
| Diisopropyl ether | C ₆ H ₁₄ O | 4 | Isopropyl ether, diisopropyl oxide, 2-isopropoxypropane |
| Dimethyl sulfate | C ₂ H ₆ SO ₄ | 4 | Dimethyl ester of sulfuric acid, methyl sulfate |
| Dimethylaniline | C ₈ H ₁₁ N | 4 | Xylidine |
| Dioxane | C ₄ H ₈ O ₂ | 4 | Diethylene dioxide, diethylene ether |
| Ethane | C ₂ H ₆ | 1 | |
| Ethanol | C ₂ H ₆ O | 4 | Ethyl alcohol |
| Ethyl acetate | C ₄ H ₈ O ₂ | 4 | Acetic acid ethyl ester, acetic ester, acetic ether, ethyl ethanoate |
| Ethyl acrylate | C ₅ H ₈ O | 4 | Ethyl propenoate, ethyl ester of acrylic acid |
| Ethyl bromide | C ₂ H ₅ Br | 4 | Bromoethane, monobromoethane |
| Ethyl chloride | C ₂ H ₅ CI | 3 | Chloroethane, hydrochloric ether |
| Ethyl formate | C ₃ H ₆ O ₂ | 3 | Ethyl methanoate, formic acid ethyl ester, formic ether |
| Ethyl mercaptan | C ₂ H ₆ S | 3 | Ethanethiol, ethyl sulfhydrate, mercaptoethane |
| Ethylbenzene | C ₈ H ₁₀ | 4 | Ethylbenzol |
| Ethylene | C ₂ H ₄ | 1 | Ethene |
| Ethylene oxide | C ₂ H ₄ O | 3 | Oxirane, epoxyethane |
| Formaldehyde | | 2 | |
| Formic acid | CH ₂ O | 2 | |
| Freons | CH ₂ O ₂ | 2 | |
| Heptane | 0.11 | 4 | N-Heptane, isoheptane, dimethylpentane |
| | C ₇ H ₁₆ | | in-i leptarie, isoneptarie, dimetriyipentarie |
| Heptene | C ₇ H ₁₄ | 4 | |
| Hexane | C ₆ H ₁₄ | 3 | N-Hexane |
| Hexene | C ₆ H ₁₂ | 2 | Butyl ethylene |
| Hydrogen | H ₂ | 1 | |
| Hydrogen bromide | HBr | 2 | |
| Hydrogen chloride | HCI | 1 | |
| Hydrogen fluoride Hydrogen iodide | HF | 2 | |
| Hydrogen selenide | HI H ₂ Se | 1 | |
| Hydrogen sulfide | | 1 | |
| Indole | H ₂ S C ₈ H ₇ N | 4 | Benzopyrrole, ketole, benzazole |
| lodine | | 4 | Delizopytiole, ketole, belizazole |
| lodoform | l ₂ | 4 | |
| Isoprene | CHI ₃ C ₅ H ₈ | 2 | 2-Methyl-1,3-butadiene |
| Lactic acid | | 4 | 2-ivietriyi-1,3-butadiene |
| Mercaptan | C ₃ H ₆ O ₃ | 4 | |
| Mercury vapours | R-SH | 1 | |
| Mesityloxide | C ₆ H ₁₀ O | 4 | 4-Methyl-3-penten-2-one, isopropylidene acetone, |
| Methane | CH ₄ | 1 | dimethylvinyl methyl ketone |
| Methanol | CH ₄ O | 3 | Methyl alcohol |
| Methyl acetate | C ₃ H ₆ O ₂ | 3 | Acetic acid, methylester, methyl ethanoate |
| Methyl acrylate | C ₄ H ₆ O ₂ | 4 | Methyl propenoate, propenoic acid ester |
| Methyl bromide | CH₃Br | 3 | Monobromomethane, bromomethane |
| Methyl butyl ketone | CH-CL | 3 | Hexanone, methyl butyl ketone, propylacetone |
| Methyl chloride Methyl ethyl ketone | CH₃CI C₄H ₈ O | 4 | Monochloromethane, chloromethane Ethyl methyl ketone, butanone, methylacetone, MEK |
| Methylcyclohexane | C ₇ H ₁₄ | 4 | Cyclohexylmethane, hexahydrotoluene, toluene hexahydride |
| Methylcyclohexanol | C ₇ H ₁₄ O | 4 | Hexahydromethylphenol, hexahydrocresol |
| Methylcyclohexanone | C ₇ H ₁₄ O | 4 | |



| Compound | Molecular formula | Capacity index | Synonyms |
|-------------------------|---|----------------|--|
| Methylene chloride | CH ₂ Cl ₂ | 4 | Dichloromethane, methylene dichloride |
| Methylether | | 3 | |
| Methylisobutylketone | C ₆ H ₁₂ O | 4 | 4-Methyl-2-pentanon, hexone, isopropylketone |
| Methylmercaptan | CH ₄ S | 4 | Thiomethane, mesh |
| Naphthalene | C ₁₀ H ₈ | 4 | Naphtalin, naphthene, tar camphor |
| Nicotine | C ₁₀ H ₁₄ N ₂ | 4 | (S)-3-[1-Methylpyrrolidin-2-y]pyridine |
| Nitric acid | HNO ₃ | 2 | Hydrogen nitrate |
| Nitrobenzene | C ₆ H ₅ NO ₂ | 4 | Nitrobenzol, oil of mirbane |
| Nitroethane | C ₂ H ₅ NO ₂ | 4 | |
| Nitrogen dioxide | NO ₂ | 1 | |
| Nitroglycerine | C ₃ H ₅ N ₃ O ₉ | 4 | Glyceryl trinitrate, propantriol-trinitrate, trinitroglycerine |
| Nitromethane | CH ₃ NO ₂ | 4 | Nitrocarbol |
| Nitropropane | C ₃ H ₇ NO ₂ | 4 | |
| Nitrotoluene | C ₇ H ₇ NO ₂ | 4 | Methylnitrobenzene |
| Nonane | C ₉ H ₂₀ | 4 | Nonyl hydride |
| Octane | C ₈ H ₁₈ | 4 | |
| Octene | C ₈ H ₁₆ | 4 | |
| Ozone | O ₃ | 4 | Triatomic oxygen |
| Palmitic acid | C ₁₆ H ₃₂ O ₂ | 4 | Hexadecanoic acid, hexadecylic acid, cetylic acid |
| P-Dichloro-benzene | C ₆ H ₄ Cl ₂ | 4 | P-DCB, dichloroxide |
| Pentane | C ₅ H ₁₂ | 3 | |
| Pentanoic acid | C ₅ H ₁₀ O ₂ | 4 | Propylacetic acid, butanecarboxylic acid |
| Pentanone | C ₅ H ₁₀ O | 4 | Ethyl acetone, methyl propyl ketone |
| Pentene | C ₅ H ₁₀ | 2 | Propylethylene |
| Pentin | C ₆ H ₁₀ | 2 | Butylacetylene |
| Phenol | C ₆ H ₅ OH | 4 | Carbolic acid, hydroxybenzene, phenyl alcohol |
| Phosgene | CCI ₂ O | 3 | Carbon oxychloride, carbonyl chloride |
| Propadiene | C ₃ H ₄ | 1 | Dimethylenemethane, dimethylenecarbon |
| Propanol | C₃H ₈ O | 4 | Isopropyl alcohol, IPA, hydroxypropane, isopropane |
| Propene | C ₃ H ₆ | 2 | Methylethene, propylene |
| Propionaldehyde | C ₃ H ₆ O | 2 | Methylacetaldehyde, propanal, propionic aldehyde |
| Propionic acid | C ₃ H ₆ O ₂ | 4 | Carboxyethane, ethanecarboxylic acid, ethylformic acid |
| Propyl acetate | C ₅ H ₁₀ O ₂ | 4 | |
| Propyl chloride | C ₃ H ₇ Cl | 4 | Chlorodimethylmethane |
| Propyl mercaptan | C₃H ₈ S | 1 | Propylthiol, propanethiol |
| Propylether | C ₆ H ₁₄ O | 4 | |
| Prussic acid | HCN | 2 | Hydrocyanic acid, hydrogen cyanide |
| Pyridine | C ₅ H ₅ N | 1 | Azabenzene, azine |
| Silicon-ethyl compounds | 0311311 | 4 | |
| Skatole | C ₉ H ₉ N | 4 | 3-Methylindole |
| Styrene | C ₈ H ₈ | 3 | Ethyl benzene, phenylethylene, vinyl benzene, styrol |
| Sulfur dioxide | SO ₂ | 1 | Sulfurous anhydride, sulfur oxide |
| Sulfur trioxide | SO ₃ | 2 | Sulfuric anhydride |
| Sulfuric acid | H ₂ SO ₄ | 4 | Oil of vitriol |
| Tetrachloroethane | C ₂ H ₂ Cl ₄ | 4 | |
| Tetrachloroethylene | C ₂ Cl ₄ | 4 | Perchloroethylene |
| Tetrachloromethane | CCL ₄ | 4 | Carbon tetrachloride |
| Toluene | C ₇ H ₈ | 4 | Methyl benzene, methyl benzol, phenyl methane, toluol |
| Trans-2-butenal | C ₄ H ₆ O | 2 | Crotonaldehyde |
| Trichloroethane | C ₂ H ₃ Cl ₃ | 4 | Ethane trichloride, vinyl trichloride |



| Compound | Molecular formula | Capacity index | Synonyms |
|-------------------|---|----------------|--|
| Trichloroethylene | C ₂ HCl ₃ | 4 | Ethylene trichloride, trichloroethene, trilene |
| Urea | CH ₄ N ₂ O | 4 | Carbamide, carbonyldiamide |
| Uric acid | C ₅ H ₄ N ₄ O ₃ | 4 | Lithic acid, trihydroxypurine |
| Valeral | C ₅ H ₁₀ O | 2 | Valeric aldehyde, amyl aldehyde, pentanal |
| Xylene | C ₈ H ₁₀ | 1 | |

ADSORPTION OF ODOURS

In addition, active carbon removes many smells and odours such as:

- · Adhesives, paint, plastic and solvent vapours
- Antiseptic vapours, anesthetic gases, disinfectants and other smells in medical settings
- Animal odours, poultry smells
- Body smells, decaying odours, burned flesh
- Cooking, burned food, burned fat and other kitchen odours
- Exhaust gases, combustion, diesel smells
- Mould smells
- Tobacco smoke odour

The capacity index for these odours is generally good, between 3 and 4.*

* Note: Odours are often complex mixtures of substances and their composition may vary. Therefore the exact capacity indexes are not given here.